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What area of study in chemistry is concerned with the heat transfers that ...
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Workbook CHAPTER
17, Thermochemistry (continued) 6. In thermochemical calculations, is the direction of heat flow given from the ...
Look at Table 17.3 on page 522 to help you answer Questions 8 and 9. 8.

SECTION 17.1 THE FLOW OF ENERGY

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HEAT AND WORK (pages 505-510)

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Workbook Chapter 9 - The Mole Concept
W9.1 (a) 1 mole of Li atoms = 7 g (b) 1 mole of Na atoms = 23 g (c) 1 mole of Ca atoms = 40 g (d) 1 mole of Fe atoms = 56 g (e) 1 mole of Ag atoms = 108 g (f) 1 mole of Pb atoms = 207 g
W9.2 (a) 1 mole of Cl atoms = 35.5 g 10 moles of Cl atoms = 35.5×10

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Atomic Structure. atom. atomic mass.
atomic mass unit (amu) atomic number
(Z) the smallest unit of an element that
maintains the properties.... the weighted
average of the masses of the isotopes of
an eleme.... a unit of mass equal to one-
twelfth

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In basic solution, $[\text{OH}^-] > 1 \times 10^{-7}$
 $\text{M} > [\text{H}^+]$. Hydrogen ion cannot appear
as a reactant because its concentration

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is essentially zero. If it were produced, it would instantly react with the excess hydroxide ion to produce water.

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synthetic p

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