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Supplemental Practice Problems 873 Practice Problems c. an atom that contains 1 electron d. an atom

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that contains 85 protons. 3. Use the periodic table to write the name and the symbol for each element identified in question 2. 4. An isotope of copper contains 29 electrons, 29 protons, and 36 neutrons. What is the mass number of this isotope? 5.

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balanced chemical  
equations in terms of  
particles, moles, and  
mass. Show that the  
law of conservation of  
mass is StoichiometryS  
toichiometry - Weebly  
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105 10 14; the answer  
will be about 20 10  
14, or 2 10 13. ...  
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15-19 page 18 24. The  
mass values of

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of moles in each of the  
following quantities. a.  
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Mol b. 346 g zinc S.2.q  
mol c. 115 g nickel 5...

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aluminum is placed in  
a 25-mL graduated  
cylinder containing

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10.0 mL of water.  
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